CBCS/B.Sc./Hons./3rd Sem./Chemistry/CEMACOR07T/2019



WEST BENGAL STATE UNIVERSITY B.Sc. Honours 3rd Semester Examination, 2019

CEMACOR07T-CHEMISTRY (CC7)

ORGANIC CHEMISTRY-III

Time Allotted: 2 Hours

IBRAF

1471



Full Marks: 40

2

2

2

3

2

2

3

The figures in the margin indicate full marks. Candidates should answer in their own words and adhere to the word limit as practicable. All symbols are of usual significance.

Answer any four questions taking one from each unit

UNIT-1

1. (a) How can you achieve the following transformations?

 $CH_3CD = CH_2 \leftarrow CH_3 - C \equiv CH \rightarrow CH_3CH = CHD$

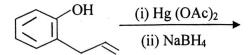
- (b) Identify the product(s) in each of the following reactions and suggest mechanism 2×2 for their formation
 - (i) (i) B_2H_6 THF (ii) $H_2O_2 - \overset{\Theta}{OH}$

(ii) $CF_3 - CH = CH_2 \xrightarrow{HBr}$

- (c) Ozonisation of 2,3-dimethylbut-2-ene in presence of formaldehyde gives the ozonide of isobutene as one of the products. Explain.
- (d) In conversion of but-2-yne to *trans*-but-2-ene by Na/liquid NH₃ and EtOH, *cis*-alkene is formed in negligible or no amount— Explain mechanistically.
- 2. (a) Write the products obtained in the reaction of E-but-2-ene with carbene generated from diazomethane in absence and presence of nitrogen. Account for the observed difference.
 - (b) How can you carry out the following conversion?

$$CH_3 - CH_2 - C \equiv CH \longrightarrow CH_3 - CH_2 - CH_2CHC$$

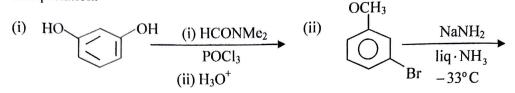
(c) Identify the product of the following reaction and suggest mechanism of the reaction:



(d) Addition of one mole HBr to buta-1,3-diene at -78°C produces 81.5% of 3-bromobut-1-ene and 18.5% of 1-bromo-but-2-ene whereas at 25°C, the yields are 44% and 56% respectively. Explain these observations.

UNIT-2

3. (a) Rationalise the fates of the following reactions in terms of mechanistic 2+2 interpretation.



(b) Predict the major and minor products of the following reaction with brief 2 explanation.

4. (a) Predict the product with mechanism

HO OH
$$+CH_3 - C \equiv N \quad \frac{(1) \text{ HCl/Zn } (CN)_2}{(2) \text{ H}_2\text{O}}$$

- (b) Explain 'Ipso Substitution' with proper example.
- (c) 2, 4, 6-trinitrochlorobenzene undergoes ready alkaline hydrolysis- Explain.

UNIT-3

- 5. (a) How can you carry out the following conversions?
 - (i) $CH_3CHO \longrightarrow CH_3CDO$

- (b) Explain the following observations.
 - (i) Both *p*-nitrobenzaldehyde and *p*-dimethylamino benzaldehyde fail to undergo benzoin condensation but a mixture of these two compounds easily responds to benzoin condensation.
 - (ii) Alkaline hydrolysis of S-1-phenylethyl acetate gives S-1-phenyl ethanol but acid hydrolysis of the same ester furnishes racemic 1-phenylethanol.
- (c) Identify compounds A and B in the following sequence of reactions and suggest 2+2+1 mechanism for their formation. Why compound B is formed as a pair of enantiomers?

2 .

PhCOCH₃
$$\xrightarrow{\text{SeO}_2}$$
 $A \xrightarrow{\text{i. NaOH}} B$
glacial $A \xrightarrow{\text{ii. NaOH}} B$
AcOH

4+3

2

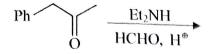
2

2

2×2

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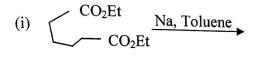
(d) Write the product of the following reaction with plausible mechanism.



- 6. (a) Compare the reactivity of acetaldehyde and acetone in aldol reaction.
 (b) Write the products of the following reaction with plausible mechanism:
 CH₃CHOHCH OHCH₃ <u>HIO₄</u>
 - (c) What happens when benzaldehyde is heated with acetic anhydride in the presence 3 of sodium acetate? Write the mechanism of the reaction.
 - (d) Can alcohol be used as a solvent in LiAlH₄ reduction of carbonyl compounds? 1+1 Explain.

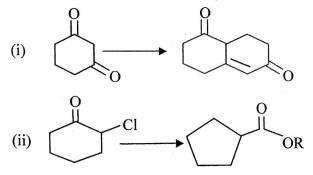
Explain the role of Li[®] in LiAlH₄ reduction of carbonyl compounds.

(e) Predict the products of the following reaction with plausible mechanism. $2 \times 2 = 4$



(ii)
$$\frac{Br}{(3) PhCH = CHCHO}$$

(f) How can you carry out following transformations?



UNIT-4

7. (a) Predict the products of the following reaction with plausible mechanism.

$$(CH_{3})_{2} CHCOCH(CH_{3})_{2} \xrightarrow{(1) (CH_{3})_{2} CHLi}?$$

$$(1) (CH_{3})_{2} CHMgBr$$

$$(2) H_{3}O^{+}?$$

 $2 \times 2 = 4$

2

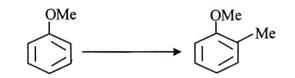
2+2

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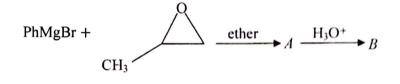
(b) How can you carry out the following conversion?



8. (a) How can you carry out the following conversion using an organo-lithium as the initial reagent?



- (b) Apply Corey House synthesis to prepare the alkane $Me_3C CH_2 CH_2 CH_3$
- (c) Identify A and B in the following reaction sequence—



2

3

1