

WEST BENGAL STATE UNIVERSITY

B.Sc. Honours 5th Semester Examination, 2021-22

CEMACOR12T-CHEMISTRY (CC12)

Time Allotted: 2 Hours Full Marks: 40

The figures in the margin indicate full marks.

Candidates should answer in their own words and adhere to the word limit as practicable.

All symbols are of usual significance.

Answer any five questions taking one from each unit

UNIT-I

- 1. (a) Outline Bogert-Cook Synthesis of Phenanthrene. How can you minimize the formation of the undesired spirocyclic product in this reaction?
 - (b) (i) How can you accomplish the following transformation? 2+2

(ii) Carry out the following transformation with plausible mechanism.

$$\stackrel{OH}{\longrightarrow} \stackrel{NH_2}{\longleftarrow}$$

- (c) (i) Compare the basicity of 2-methyl indole and 3-methyl indole.
- $1\frac{1}{2} + 1\frac{1}{2}$
- (ii) State, with mechanism, how can you convert benzaldehyde into 1-methylisoquinoline.
- 2. (a) What happens when 2-naphthol is treated with ferric chloride?

1

(b) Predict the product(s) in the following reactions and suggest mechanism in each $2\frac{1}{2} + 2\frac{1}{2}$ case.

i.
$$\frac{\text{HCONMe}_2}{\text{POCl}_3}$$
 ii. $\frac{\oplus}{\text{H}}$

- (c) Furan reacts differently with nitronium fluoroborate and acetyl nitrate (in pyridine) to give 2-nitrofuran. Explain.
- _

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(d) How 2-bromonapthalene is prepared from naphthalene?

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UNIT-II

3. (a) Predict the product(s) and give mechanism for the following reaction.

Me₂N OTs (i) Et₃N/EtOH/ Δ \oplus (ii) H/H₂O

 $1\frac{1}{2}$

2

 $2\frac{1}{2}$

3

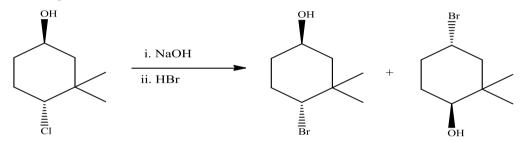
 $1\frac{1}{2}$

 $1\frac{1}{2}$

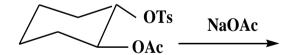
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- (b) What happens when *cis*-1,3-cyclohexane dicarboxylic acid is heated? Comment on the chirality of the product.
- (c) Explain the fact that *trans*-4-*tert*-butylcyclohexyl tosylate undergoes bimolecular reaction with the base bromide and thiophenolate, although not with the much stronger base ethoxide.
- 4. (a) Provide a mechanistic rationalisation to explain the stereochemical aspects of the following reaction.



- (b) Write the product with proper stereochemistry for the following reaction and explain:
 - NH₂
 H NaNO₂/HCl
- (c) Predict the product of the following reaction with suitable mechanism:



UNIT-III

- 5. (a) In the thermal ring opening of *trans*-3,4-dimethylcyclobutene, two products can be formed by conrotatory mode, but only one is actually formed. Identify the possible products. Which one is observed and why?
 - (b) Propose a mechanism for the following reaction clearly indicating the pericyclic steps involved therein.

5059

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(c) Predict the product with stereochemistry in the following reaction and indicate the reaction pathways:

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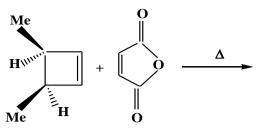
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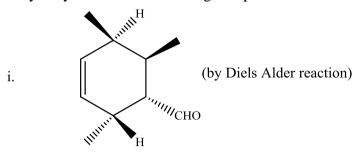
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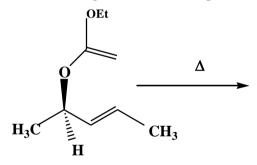
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6. (a) How can you synthesise the following compounds as directed?



(b) Predict the product of the following reaction with explanation:



(c) Thermal 1,3-sigmatropic shift of hydrogen is symmetry forbidden but thermal 1,3-sigmatropic shift of an alkyl group may be symmetry allowed. Explain in terms of FMO theory.

UNIT-IV

- 7. (a) β -D-Glucopyranose is oxidized at 250 times faster rate than α -D-Glucopyranose using Br₂/H₂O. Explain.
 - (b) How can you convert D-glucose into:(i) meso tartaric acid and (ii) D-glucuronic acid?
 - (c) D-Glucose and another aldohexose (A) give the same product when treated with sodium amalgam. Find out the structure of (A). To which family between D and L, does (A) belong?
 - (d) How would you distinguish chemically between ribose and 2-deoxyribose?

5059 Turn Over

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8. (a) How many moles of HIO₄ will be required for the oxidation of one mole of the following compound? Write down the products.

CH(OCH₃)
(CHOH)₃
CH

CH₂OH

(b) Why osazone formation doesn't proceed beyond C-2 of an aldohexose?

2

(c) Clearly represent the most stable conformation of the β -pyranose form of the following sugars and justify.

2

2

(i) β -D-allopyranose (ii) β -L-glucopyranose

(d) Diisopropylidene derivative of D-glucose can be O-methylated at C-3, but that of D-galactose cannot give the same result. — Explain.

2

UNIT-V

9. (a) Specific rotation of an amino acid is pH-dependent. — Justify.

2 1+2

(b) Name one amino acid that produces yellow colour with ninhydrin. Write down the pertinent reaction.

3

(c) A tripeptide X on hydrolysis gives two amino acids, Glu (2 equivalent) and Ala (1 equivalent). X does not react with 2,4-dinitrofluorobenzene. Ala is released first when X is incubated with carboxypeptidase. Deduce the structure for X.

2

(d) Define the isoelectric point of an amino acid. How can lysine (pI 9.6) be separated from glycine (pI 5.97) by electrophoresis?

1+2

10.(a) Explain the role of cellular water in the stabilisation of a DNA duplex. Show the G-C base pairing in DNA.

1 1 2

(b) In deionised water, isoelectric and isoionic points of an amino acid are identical. — Why?

7

(c) Trace the route of synthesis of Ph-CH(NH₂)-CO₂H from phthalimide.

2

1

(d) Predict the product with mechanism when leucine is heated with acetic anhydride in presence of pyridine.

2

(e) How can you synthesise L-tryptophan using azlactone method?

2

N.B.: Students have to complete submission of their Answer Scripts through E-mail / Whatsapp to their own respective colleges on the same day / date of examination within 1 hour after end of exam. University / College authorities will not be held responsible for wrong submission (at in proper address). Students are strongly advised not to submit multiple copies of the same answer script.

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5059 4